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Great Picnic Mix Up:
Crash Course Kids~~

~~#19.1 Mixtures vs~~

Solutions | Know the
Difference MIXTURES

AND THEIR

CHARACTERISTICS |

Homogeneous and

Heterogeneous

Mixture | Science 6 |

by Sir C.G. What is a

solution? | Solutions |

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~~EXPLORE ACTIVITY~~

~~5.5 CD: MIXTURES~~

~~AND SOLUTIONS~~

~~(Grade Level 5)~~

~~Mixtures vs. Solutions~~

~~Homogeneous and~~

~~Heterogeneous~~

~~Mixtures Examples,~~

~~Classification of~~

~~Matter, Chemistry 5th~~

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Mixtures and

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Overview Pure

Substances and

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(Classification of

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Compounds, and

Mixtures Mixtures -

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...

A solution is a
mixture in which the
particles of a
different substance
are evenly

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distributed, and are too small to see with the naked eye. A solution appears to be a single substance. The sugar and water make a solution. The sugar crystals dissolve in the water, changing the physical properties of the liquid.

Mixtures and

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Learning ...

Mixtures are a physical combination of two or more substances. Mixtures can be classified as heterogeneous or homogeneous, a type of heterogeneous mixture is a salad. A type of homogenous mixture is lemonade.

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In a mixture each substance keeps its properties, like if you put avocado in a salad and took it out it would still be avocado.

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Mixtures And

Solutions

if the particles in a mixture are the size of atoms or

molecules. this

happens when one

substance dissolves

in another, like sugar

in water. solution.

solvent. a part in a

solution that does

the dissolving. solute.

a part in the solution

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Science gets dissolved.

solubility, the
greatest amount of
solute that a given
amount of solvent
can dissolve in.

Physical Science:

Mixtures & Solutions

Flashcards | Quizlet

saturation: A solution
that has reached its
maximum

concentration. solute:

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The solid, liquid, or gas that dissolves in the liquid of a solution. The salt is the solute in a salt-water solution.

solution: A

homogenous mixture formed by the dissolution of a liquid, solid or gas in a liquid.

Properties of

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Mixtures vs.

Solutions: Mix It Up! -
Lesson ...

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