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Grade 9 Chemistry, Lesson 7 - The Periodic Table Part 2 - Patterns in the Table

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At that point, the reaction had finished. 2 a i Graph showing amount of product increasing with time. ii. The graph showing the amount of reactants with time. b Answer The gradient of the line on the graph of concentration versus time give the information of the rate of a reaction.

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A coffee-cup calorimeter contains 50.0 mL of distilled water at 22.7 °C. Solid ammonium bromide (3.14 g) is added and the solution is stirred, giving a final temperature of 20.3 °C. Using the same assumptions as in Example 9, find H_{soln} for NH 4 Br (in kilojoules per mole). Answer: 16.6 kJ/mol