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2 Section A Answer all questions. Write your answers in the boxes provided. 1. Ethanedioic acid is a diprotic acid. A student determined the value of x in the formula of hydrated ethanedioic acid, $\text{HOOC} \cdot \text{COOH} \cdot x\text{H}_2\text{O}$, by titrating a known mass of the acid with a $0.100 \text{ mol dm}^{-3}$ solution of $\text{NaOH} (\text{aq})$. 0.795 g of ethanedioic acid was dissolved in distilled water and made up to a total ...

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5. a ratio of electrons : aluminium ions 3 : 1 amount Al $13 \frac{1}{1.296} 10 \text{ C } 96500 \text{ C mol } 3 \llcorner \llcorner 4.48 107 \llcorner \llcorner \text{mol} \llcorner \llcorner \text{mass Al} \llcorner 4.48 107 \text{ mol } 26.98 \text{ g mol } 1 \llcorner 1.21 109 \llcorner \llcorner \text{g} \llcorner \llcorner \text{Award [3] for correct final answer. 3 5.}$

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Many students struggle with IB Chemistry SL/HL. There are so many subjects to learn, and you cannot fall behind. You need to be learning during the school year in order to ace the IB Chemistry papers. Common errors students make when studying are: #1: Avoiding the topics you didn't fully comprehend in class.

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