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6 of the chemistry textbook
and they have to examine a
lot of concepts related to
thermodynamics. They will be
learning about the
interchange between heat and
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Clapeyron equation: $dP_i^{sat} = \frac{h_i^{vap}}{RT^2} dT$
or $\ln P_i^{sat} - \ln P_i^{sat}(101 \text{ kPa}) = -\frac{h_i^{vap}}{R} \left(\frac{1}{T} - \frac{1}{373 \text{ [K]}} \right)$ so
 $P_i^{sat} = 101 \dots$

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helps in defining the
process and also the
direction of the processes.
It tells about the
possibility of a process or
the restriction of a
process. It helps in
analyzing the performance of
processes, cycles, devices
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