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(c) The potential energy of the wooden log was converted into 1.00 J of heat. (d) Since the ash weighs less than the wooden log, matter was converted directly into energy. (e)

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The burning of the 1.00 kg wooden log produced 0.96 kg of gases and 0.04 kg of ash.

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The first ionization energy for a neutral atom of chlorine is 1.25 MJ mol^{-1} and the first ionization energy for a neutral atom of argon is 1.52 MJ mol^{-1} How would the first ionization energy value for a neutral atom of potassium compare to those values?

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